

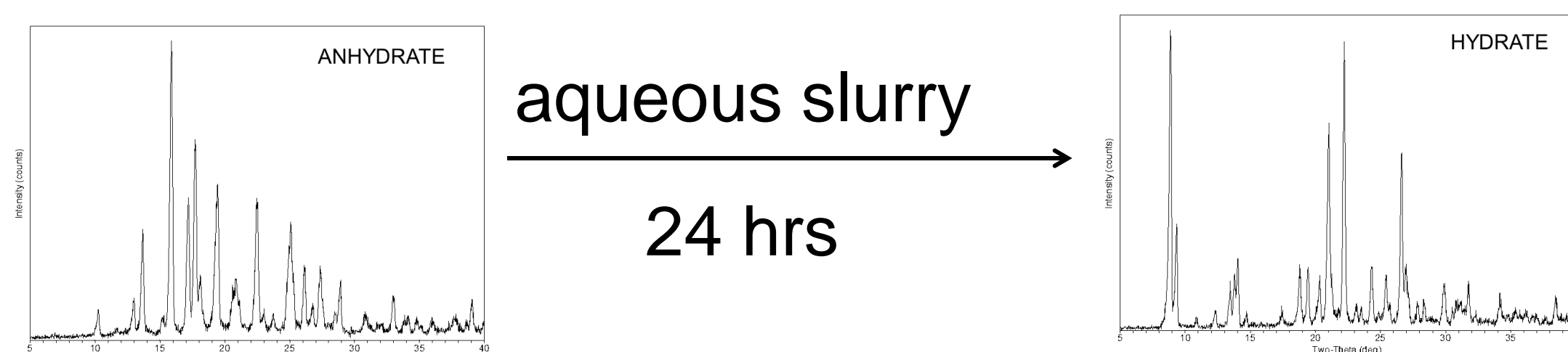
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Assessment of the nature of disorder and its effect on recrystallization tendency in a hydrate-anhydrate system

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BACKGROUND

- Investigational compound GENE, is characterized as anhydrate. Anhydrate converts to trihydrate in aqueous slurry.



- Hydrate dehydrates to form a “dehydrated hydrate/DH” which is X-ray amorphous. DH crystallizes to anhydrate at $\geq 192^\circ\text{C}$.

OBJECTIVE

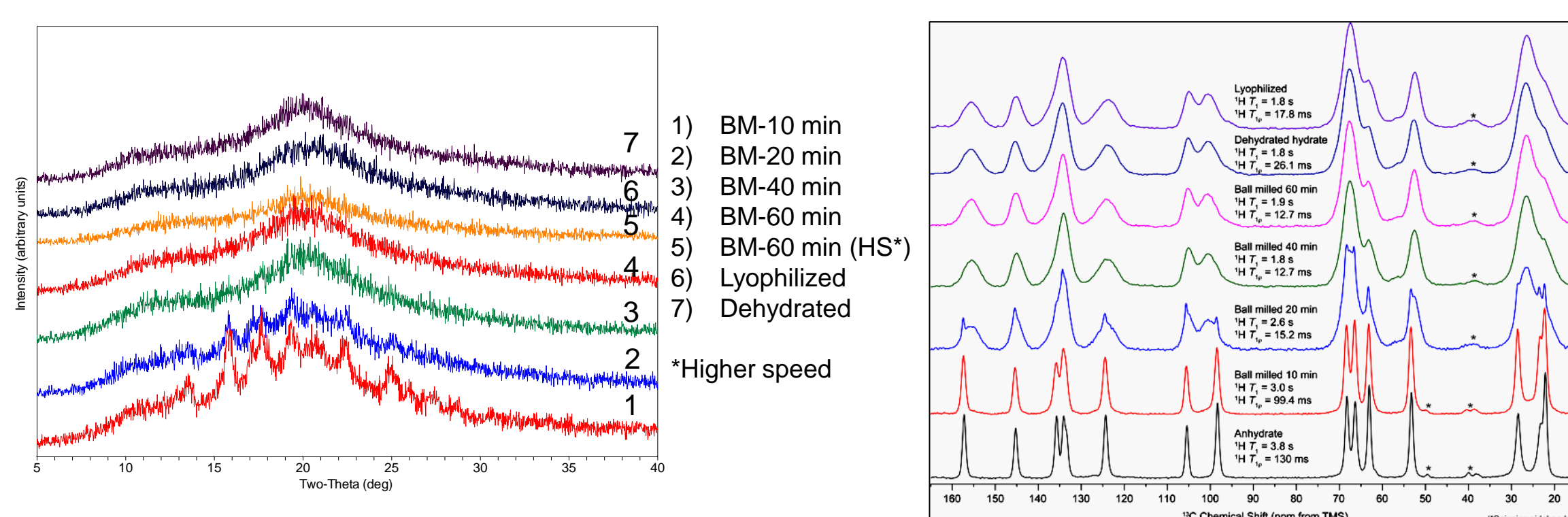
- Investigate the nature of disorder/local order in a “disordered phase” generated through different routes
- Understand effects of disorder/ “existing order” on recrystallization tendency and nature of resulting phase

METHODS

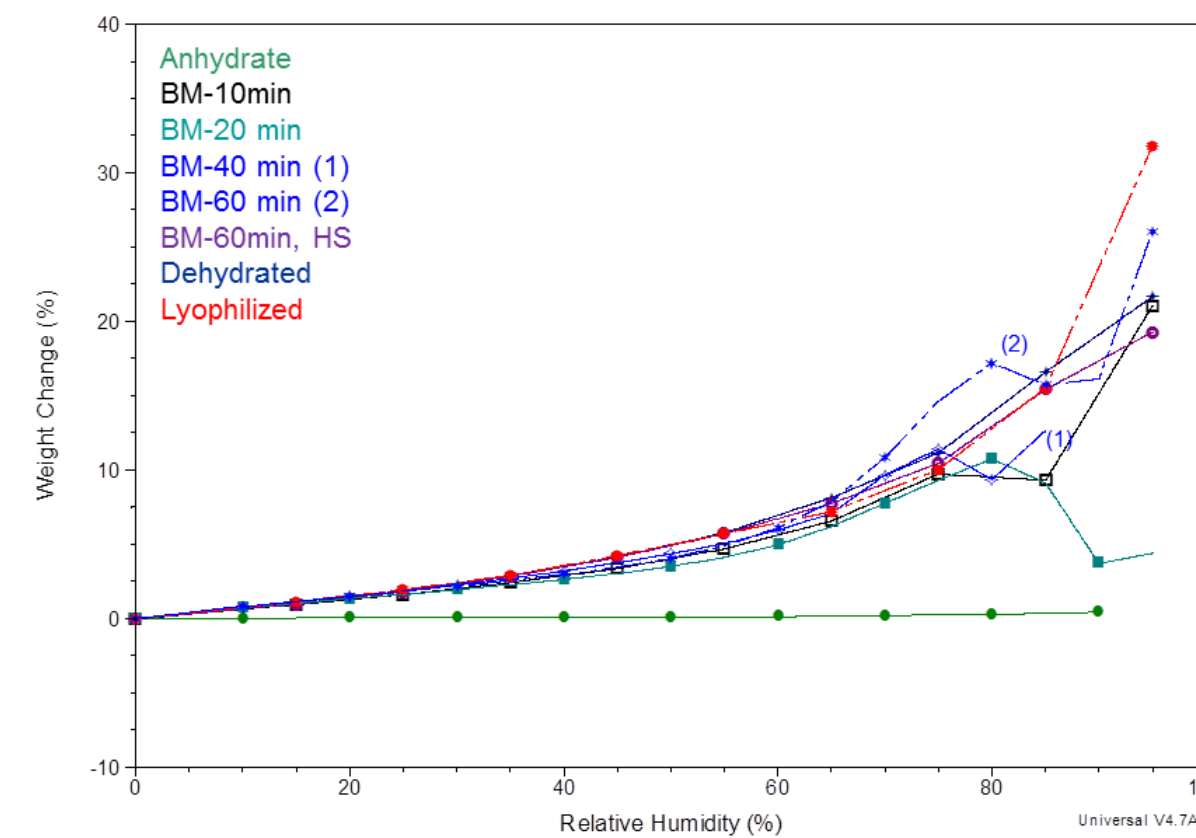
- Lattice disorder was induced by ball milling, lyophilization and dehydration
- Characterization: DSC, TGA, XRPD, SSNMR, Water sorption, computational analysis

RESULTS

Disordered phases

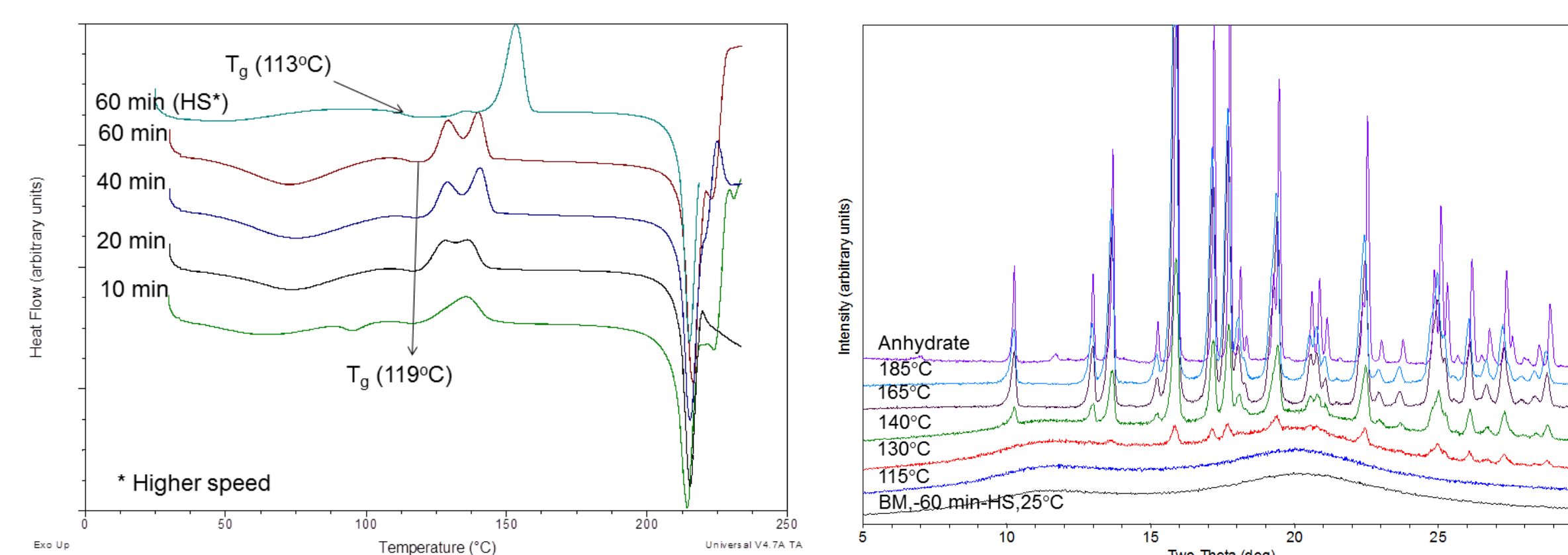


Overlaid patterns (XRPD, left) and spectra (SSNMR, right)



Water sorption (0-95% RH, 25°C)

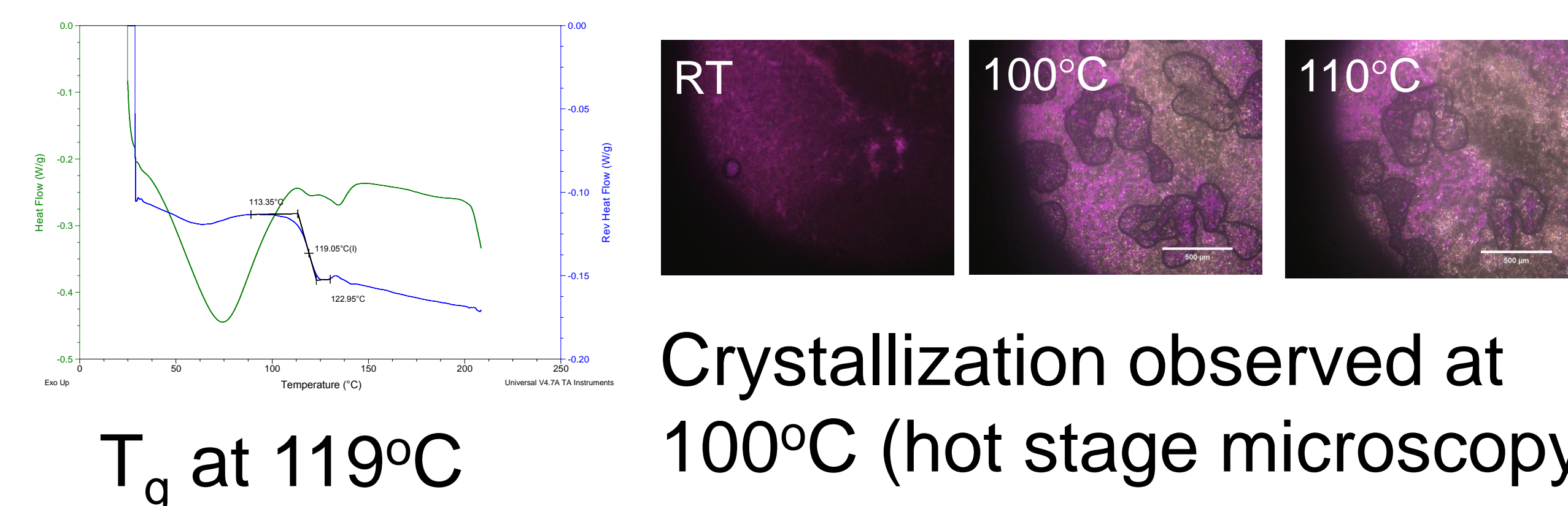
Ball Milled



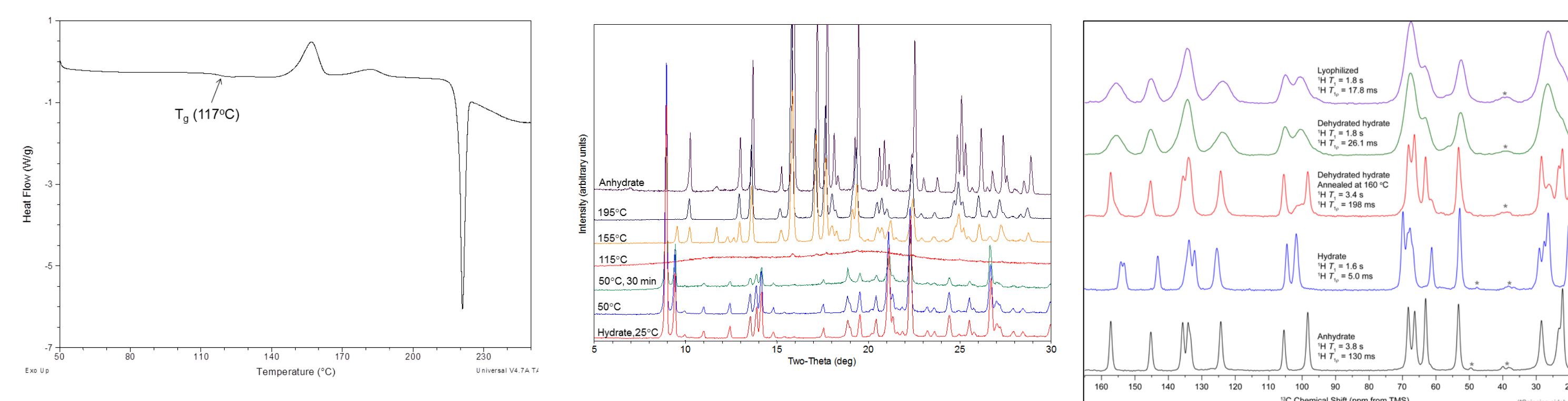
DSC traces (left) of BM samples. Crystallization exotherms overlap with T_g except for the sample ball milled for 60 min, (higher speed/HS)

VTXRPD (right) of the BM-60 min (HS) sample. Crystallization to anhydrate starts at 125°C

Lyophilized

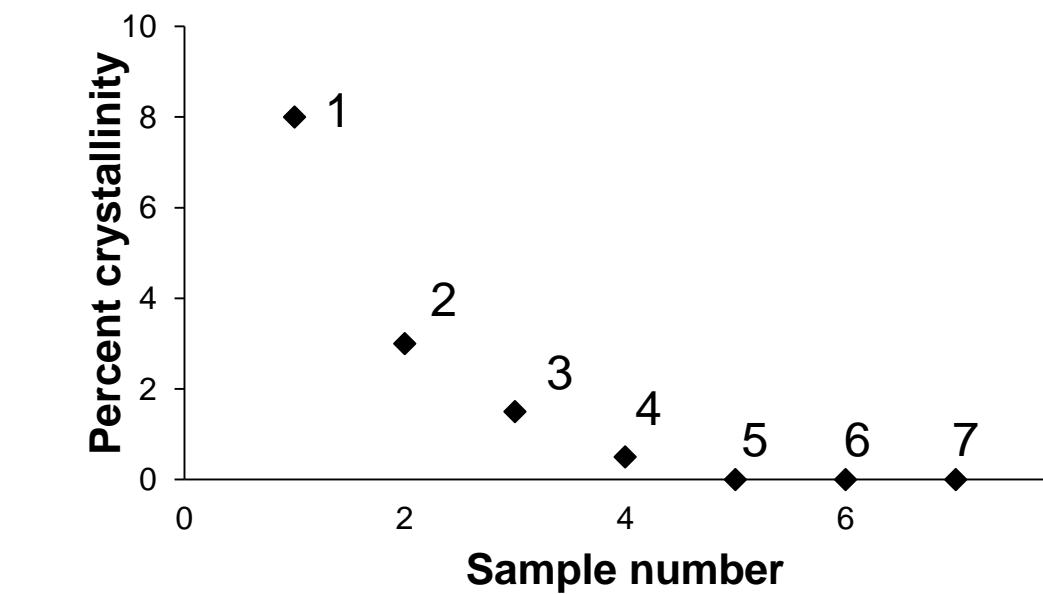


Dehydrated (DH)

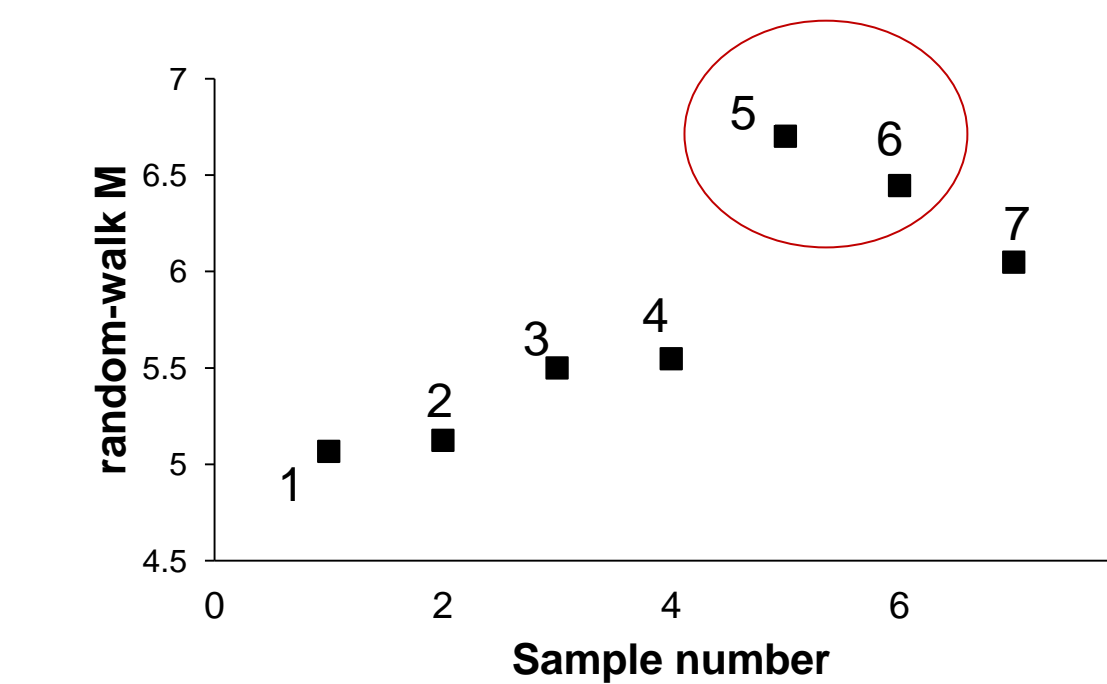


DSC (left) shows 2 crystallization exotherms for DH. VTXRPD (center) and SSNMR (right) reveal an “intermediate” phase at 155-160°C (1st DSC exotherm) which rearranges to form the anhydrate at 195°C

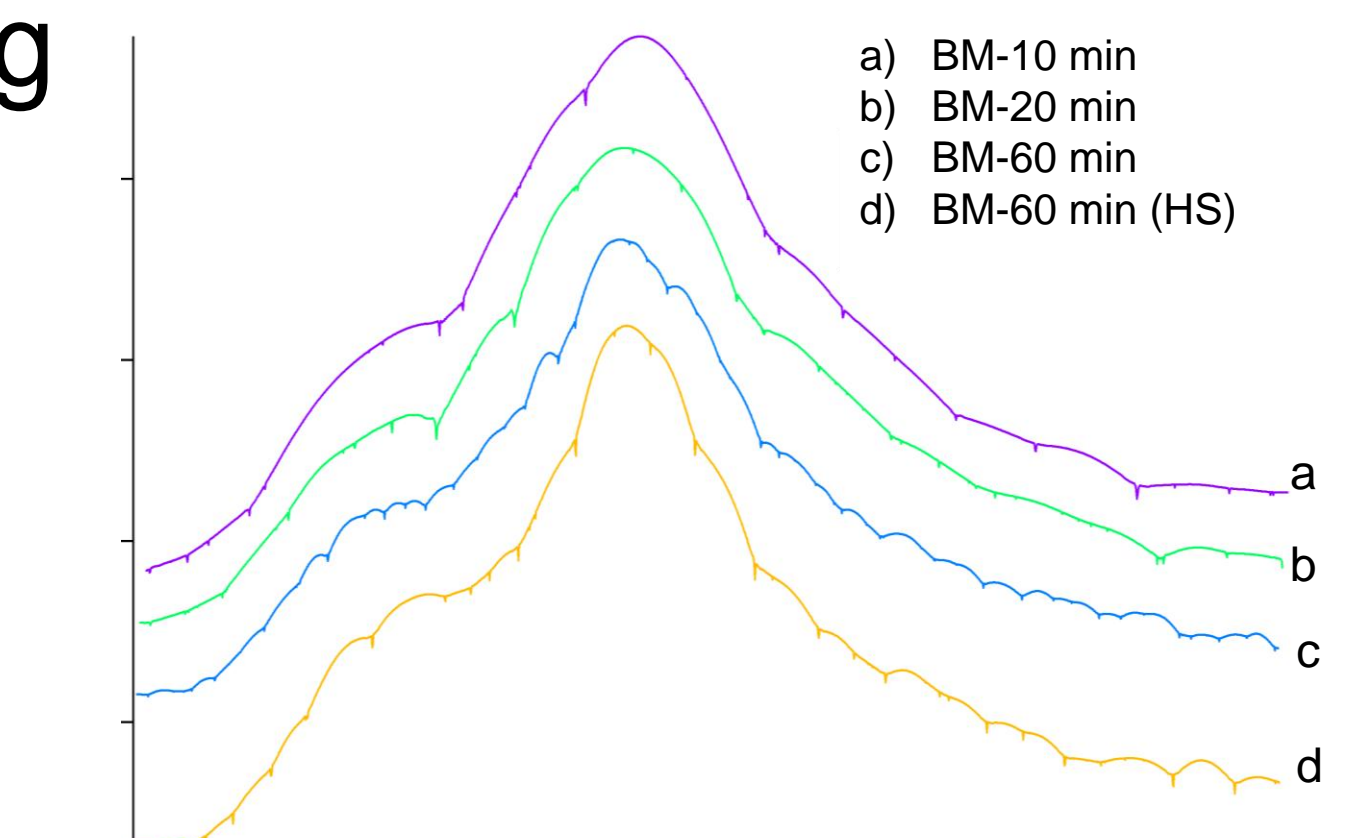
Computational analysis



Sample %crystallinity determined from digital filtering



The change in the random-walk variable M resulting from random-walk direct analysis. The increase in the value of M corresponds to the sharper primary halo profile and increased local order.



Result of applying a digital low pass filter to raw data. Diffuse scattering profile is changing under increase milling time, with the BM-60 min (HS) sample having the narrower diffuse halos.

SUMMARY

- Two crystallization endotherms observed in dehydrated and BM samples (≥ 20 min milling)
- Increase in milling time leads to greater local order. Highest degree of local order observed in ball milled (60 min, HF) and dehydrated samples.
- Since nature of local order influences propensity of crystallization and the resulting crystalline phase, the delayed onset of crystallization in these samples with greater local order suggests presence of a lattice order unlike that in the known crystalline phases

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